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Common) • Ion-dipole Forces – Present In Solutions Of Ionic Compounds In Polar Solvents Such As H<sub>2</sub>O – Hydration (solvation) • The Water Dipoles Pull The Ions Away From The Ionic Crystal And Surround Them (typical Coord. Numbers 4 Or 6) • There Is A Short Range Order Around The Ions Consisting Of H-bonded Shells Of Water Molecules Feb 4th, 2022

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Figure 11.7 Solutions Of Nonelectrolytes Such As Ethanol Do Not Contain Dissolved Ions And Cannot Conduct Electricity. Solutions Of Electrolytes Contain Ions That Permit The Passage Of Electricity. The Conductivity Of An Electrolyte Solution Is Related To The Strengt Mar 12th, 2022

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